





Elements:Carbon, Hydrogen, Oxygen (carbon & water)

	Name	Structure	How used in organism
Monosac- charides	Simple Sugars Glucose fructose galactose	All can be carbon chains or rings, $C_6H_{12}O_6$	cellular energy
Disaccha- rides	Sucrose Lactose Maltose	Formed by condensation $\downarrow_{n}^{(A)}$ $\downarrow_{n}^{(A)}$ \downarrow_{n	Broken down into single units to for energy
Polysaccharides	Starch Glycogen Cellulose &Chitin	Long chains of glucose & branched chains of glucose Densely branched chains of glucose Long straight chain of glucose, chitin has branches of nitrogen	Energy storage for plants Energy storage for animal & fungi Structural-cell walls, fiber, wood

Elements: Carbon, Hydrogen and much less Oxygen

T	RIGLYCI	How used in organism	
(F	Glycerol Polar head)	Fatty Acids (Non-Polar tails)	Lipids include oils, waxes and sterols
ner T3.html	н — с — о —	O H H H H H Saturated Fat II I I I I I No double bonds -C-C-C-C-C-H No double bonds I I I I I Animal fat, Coconut oil H H H H H Solid at room temp.	Insulation, hor- mones, protective
ilk-facts.com/fat_prin	н — с —о- 	OHHHHUnsaturated FatIIIIII $-C-C-C-C-C-H$ One double bondIIHHHHHHHHLiquid at room temp.	term energy storage
http://www.raw-m.	н — с — о- н	OHHHPolyunsaturated FatIIIIII $-C-C=C-C=C-H$ More than one double bondIIPlant Oils-walnut, corn sun-HHflower	

Sample

Macromolecules

This is the inside of the foldable

Elements: Carbon, Hydrogen. Oxygen and Nitrogen

	Name	Structure	How used in organism
monomer	Amino acids There are 22 natu- rally occurring amino acids	http://www.hcc. <u>mmscu.edu/</u> <u>chem/V.27/</u> <u>page id 17100</u> <u>html</u> <u>Amino</u> <u>Group</u> <u>Side Chain</u> <u>Carboxylic Acid</u> <u>Group</u> <u>Carboxylic Acid</u> <u>Group</u>	Building block of proteins
Polype	Polypeptide Chain (amino acid sequence)	Peptide bonds join 2 amino acids through condensation. Poly- peptide chains twist because of hydrogen bonds. <u>Imit/ien witipedia org/wiki/</u> <u>Amino witipedia org/wiki/</u>	Support/Structural-cartilage, skin, ligaments Movement-muscles Enzymes-biological catalysts Defense-pigments, antibodies
ptide	Protein Structure	More than one polypeptide chain, twisted and folded together. http://upload.wikimedia.org/ wikipedia.commoms/6/60/ Myoglobin.png	

Elements: Carbon, Hydrogen, Oxygen, Nitrogen and Phosphorus

	Name	Structure	How used in organism
Monomer	Nucleotides	Nitrogen Bue Processar Bre Carbon Sugar, a phosphate group and a nitrogenous base <u>http://saff.jccc.edu/pdccell//biochemistry/mucleotides.html</u>	Building block of nucleic acids (found in the nu- cleus of cells)
	RNA 5-carbon sugar is ribose	A single nucleotide chain	Translates the genetic codes into proteins
Polymers	DNA 5-carbon sugar is deoxyribose	 Two nucleotide chains form the Double Helix Backbone consists of the phosphorus & ribose group Hydrogen bonds hold the nitrogen bases to form the "rungs" of the ladder 	Found in the nucleus Genetic code Regulates the activates of the cell

Alternative to Foldable

If you prefer to have more space for the students to draw. Try a Layered tent fold with 3 sheets of paper.

